

ent-9(8→15αH)abeo-17-Norkaur-8(14)-en-16-one

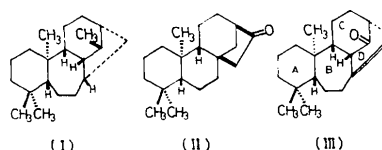
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Abstract. C₁₉H₂₈O, orthorhombic, $P2_12_12_1$, $a = 23.69$ (4), $b = 8.058$ (12), $c = 8.132$ (12) Å, $Z = 4$. The two six-membered rings are in the chair form, and the five-membered ring is in an envelope conformation. All the rings in the molecule are somewhat distorted. Repulsion between the 1,3 diaxial methyl groups attached to ring A results in rotation of the methyl H atoms from the usual staggered position.

also stereochemical evidence for the large strain expected in the structure.



Introduction. Two of the authors (MO and EF) carried out a synthesis of *ent*-9(8→15αH)abeo-kaurane (I). In this synthesis, *ent*-17-norkauran-16-one (II) was allowed to react with thallium trinitrate to afford a rearranged product which was assigned structure III on the basis of spectral and chemical evidence (Fujita & Ochiai, 1976). The present X-ray analysis of compound III provides a confirmation of the structure and

Intensity data were collected on a Rigaku automatic four-circle diffractometer using Cu $K\alpha$ radiation. The structure was solved by trial and error methods based on Patterson maps and E maps derived by use of the program *MULTAN* (Germain, Main & Woolfson, 1971). Successive cycles of anisotropic block-diagonal least-squares refinement gave an R value of 0.043 for 1571 observed reflexions. H atoms were located from

Table 1. Fractional atomic coordinates and anisotropic thermal parameters ($\times 10^4$) of the non-hydrogen atoms with estimated standard deviations in parentheses

The temperature factor expression is $T = \exp [-(h^2\beta_{11} + k^2\beta_{22} + l^2\beta_{33} + 2hk\beta_{12} + 2hl\beta_{13} + 2kl\beta_{23})]$.

	x	y	z	β_{11}	β_{22}	β_{33}	β_{12}	β_{13}	β_{23}
C(1)	1630 (2)	1034 (6)	4036 (6)	23 (1)	121 (8)	125 (8)	3 (3)	-7 (3)	21 (8)
C(2)	1883 (2)	2276 (7)	5264 (6)	34 (1)	164 (9)	121 (8)	7 (3)	-17 (3)	23 (9)
C(3)	1527 (2)	3865 (6)	5297 (6)	26 (1)	163 (9)	136 (8)	-4 (3)	2 (3)	18 (9)
C(4)	1455 (2)	4698 (6)	3613 (6)	15 (1)	144 (8)	156 (8)	-6 (2)	-5 (2)	-1 (8)
C(5)	1258 (2)	3377 (5)	2318 (5)	12 (1)	104 (7)	133 (7)	-5 (2)	-5 (2)	21 (7)
C(6)	1158 (2)	4168 (6)	609 (6)	20 (1)	125 (8)	159 (8)	-7 (2)	-18 (3)	39 (8)
C(7)	575 (2)	3654 (6)	-132 (6)	18 (1)	132 (8)	190 (9)	2 (2)	-22 (3)	14 (8)
C(8)	552 (2)	1821 (5)	-441 (6)	11 (1)	122 (7)	176 (9)	1 (2)	-11 (2)	21 (8)
C(9)	1315 (2)	337 (5)	1183 (5)	11 (1)	111 (7)	126 (7)	0 (2)	-2 (2)	9 (7)
C(10)	1606 (2)	1727 (5)	2263 (5)	11 (1)	119 (7)	116 (7)	0 (2)	-1 (2)	18 (7)
C(11)	1583 (2)	43 (6)	-534 (6)	14 (1)	165 (9)	128 (8)	3 (2)	4 (2)	-14 (8)
C(12)	1256 (2)	-1201 (6)	-1614 (6)	18 (1)	160 (9)	152 (8)	7 (3)	1 (2)	-8 (8)
C(13)	612 (2)	-816 (6)	-1644 (6)	17 (1)	150 (9)	165 (9)	-6 (3)	-11 (3)	-12 (8)
C(14)	516 (2)	1028 (6)	-1871 (6)	16 (1)	163 (9)	159 (9)	-1 (3)	-13 (2)	16 (8)
C(15)	659 (2)	598 (5)	951 (6)	12 (1)	112 (7)	141 (8)	-3 (2)	1 (2)	10 (7)
C(16)	438 (2)	-981 (5)	148 (6)	12 (1)	119 (7)	190 (9)	0 (2)	-6 (2)	2 (8)
O(17)	183 (1)	-2095 (4)	789 (5)	19 (1)	147 (6)	245 (7)	-14 (2)	2 (2)	16 (6)
C(18)	981 (2)	5996 (6)	3814 (7)	22 (1)	148 (9)	261 (12)	4 (3)	-12 (3)	-39 (10)
C(19)	1998 (2)	5649 (6)	3134 (7)	21 (1)	180 (10)	214 (11)	-25 (3)	-12 (3)	-10 (10)
C(20)	2220 (2)	1965 (7)	1659 (6)	11 (1)	204 (10)	185 (9)	-5 (2)	1 (2)	-13 (9)

Table 3. *Least-squares planes and atomic deviations*

The planes are defined by $Ax + By + Cz + D = 0$ where x, y and z are in Å along the axes a, b and c . An asterisk indicates an atom not used in the plane calculation.

$A = -0.8634$
 $B = 0.4616$
 $C = -0.2037$
 $D = 1.2841$

$A = -0.9905$
 $B = -0.0654$
 $C = 0.1208$
 $D = 1.4961$

Atom deviations (Å)

C(7)*	1.489
C(8)*	0.905
C(13)	0.001
C(14)*	0.920
C(15)	0.001
C(16)	-0.002
O(17)	0.001
H(14)*	1.372

Atom deviations (Å)

C(7)	-0.059
C(8)	0.061
C(13)	-0.059
C(14)	0.046
C(15)	0.011
C(16)*	0.534
O(17)*	1.255
H(14)*	-0.008

The single bond C(7)—C(8) (1.499 Å) is shortened by resonance with the adjacent double bond C(8)—C(14). Five-membered ring *D* has an envelope conformation. The C(16) atom deviates from the plane of the other four atoms. The least-squares plane through C(13), C(15), C(16) and O(17) makes a dihedral angle of 33.8° with that through C(7), C(8), C(13), C(14) and C(15) (Table 3). The strain in ring *D* is evidenced by the fact that the internal bond angles at C(8), C(14) and C(16) are much smaller than the normal sp^2 -hybridization angle (120°). Repulsion between the 1,3 diaxial methyl groups attached to ring *A* causes the significantly greater bond angles 114.5° for C(5)—C(4)—C(19) and

113.5° for C(5)—C(10)—C(20), resulting in the C(19)—C(20) distance of 3.24 Å. A similar structural feature has been observed in (–)-kaur-15-en-19-al (Karle, 1972) and in phyllocladan-15-yl bromoacetate (Brown & Hall, 1976). In the present case, the three methyl H atoms on C(19) are rotated about the C(4)—C(19) bond from the staggered position, while the methyl H atoms on C(20) are in the usual staggered position. The shortest H—H approach between these methyl groups is 2.08 Å.

There are no intermolecular approaches likely to influence the molecular structure.

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